

# Assessing the Role of Turbulent Mixing on Phytoplankton Dynamics in Piedmont Reservoirs

Tarek N. Aziz, Daniel Obenour, Jeremy Smithheart  
Yue Han, Alexandre Mangot, Robyn Smyth  
NC State University  
Department of Civil, Construction, and Environmental Engineering

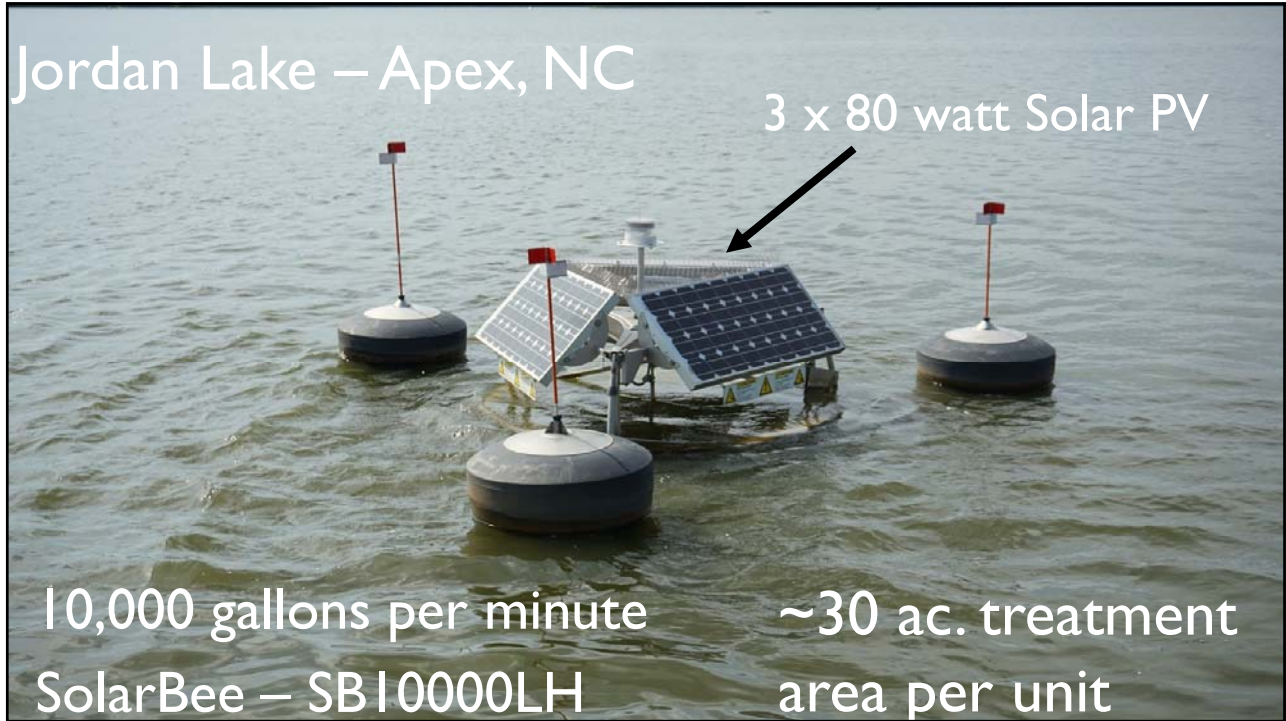
NC STATE UNIVERSITY



NC STATE UNIVERSITY

## Harmful Algal Blooms – why do we care?

- Beneficial uses of reservoir
- Taste and odor in drinking water
- Toxin production
- Disinfection byproducts



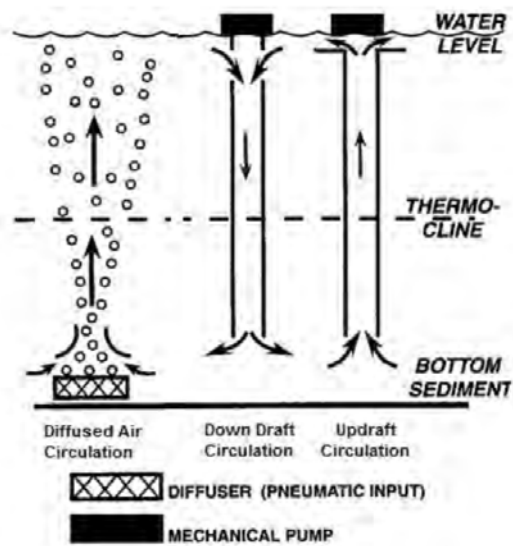


Source: Courtesy of WEARS

Figure 3.6 Views of components of the WEARS ResMix DDP system

Wagner, K. J. *Water Supply Reservoir Management Oxygenation and Circulation to Aid Water Supply Reservoir Management*; Denver, CO, 2015.

## Circulation



Wagner, K. J. *Water Supply Reservoir Management Oxygenation and Circulation to Aid Water Supply Reservoir Management*; Denver, CO, 2015.

## Mixing – A Tale of Two Lakes

- Solar Circulators – 240W solar PV per circulator (32 circulators)

240W x 32 circulators x 12 hrs sunlight x 365 days/yr

**~34,000 kWh/yr (power delivered) (very favorable calculation)**

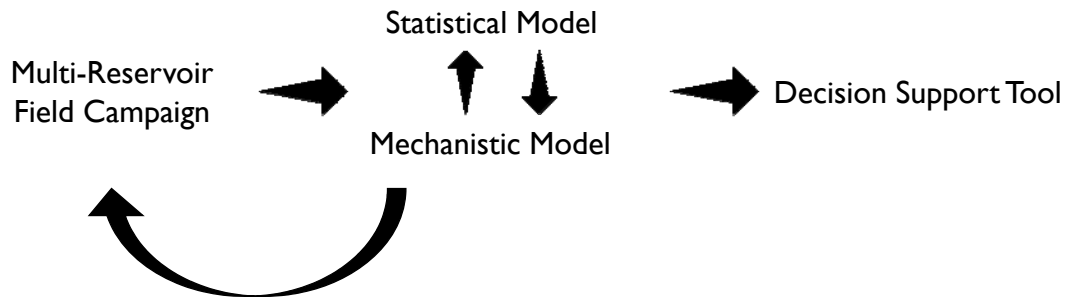
- City Lake Aeration – (from inquiry to City of High Point)

52,618 kWh/month → **631,416 kWh/yr (delivered)**

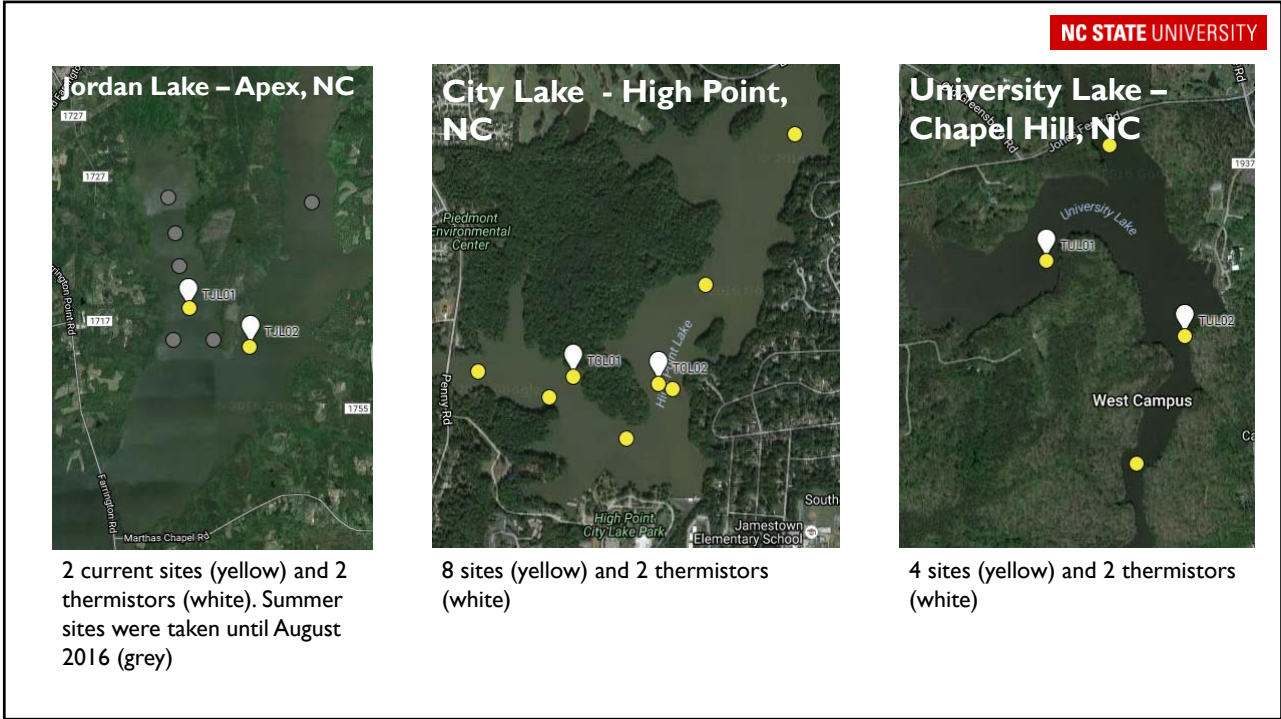
## Big Questions:

- How does mixing (natural and artificial) affect phytoplankton community structure?
  - **What about toxin production? – Astrid Schnetzer**
- What levels of mixing are needed (and when) to suppress HAB formation?
- Can we predict bloom formation or phytoplankton community composition based on mixing?

# WRRI Grant Study Objectives





## Field Campaign

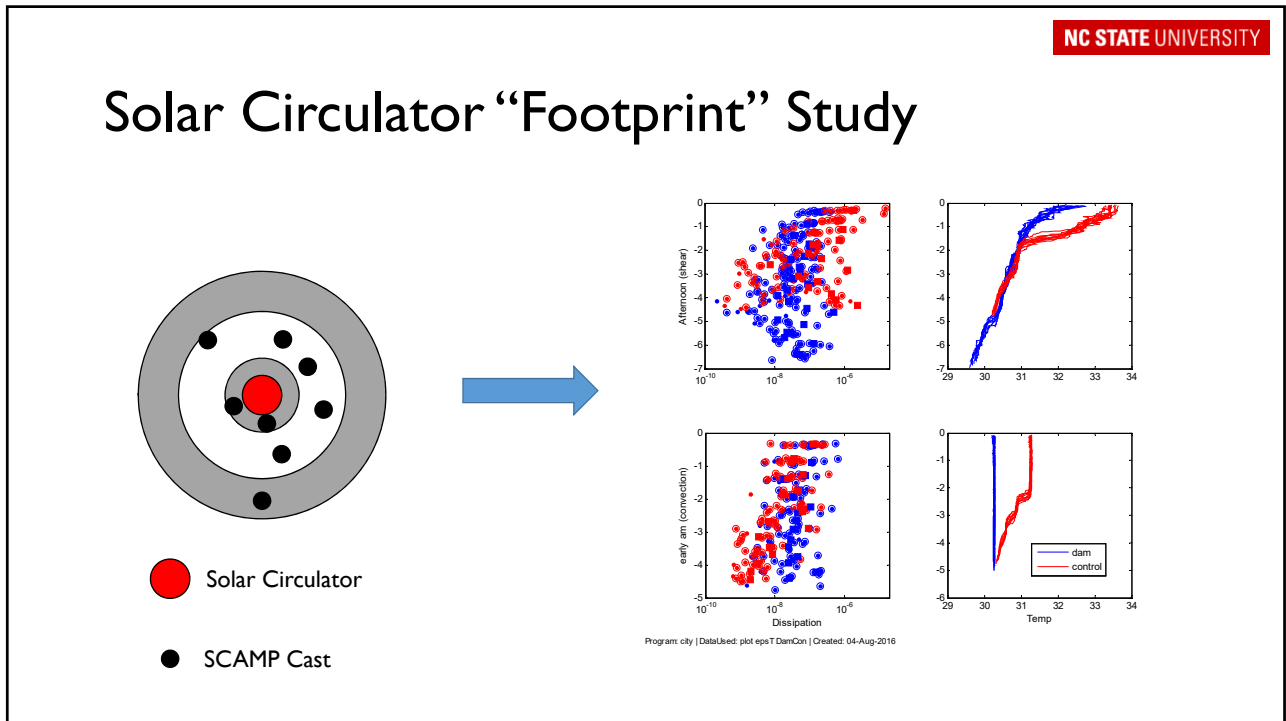


## NC STATE UNIVERSITY

# Field Campaign

Data Set	2015	2016
Jordan Lake	✓	✓
University Lake		✓
High Point City Lake		✓
Monthly Sonde Data (Temperature, LDO, pH, Conductivity, Chlorophyll Fluorescence, and Phycocyanin Fluorescence)	✓	✓
Hourly Temperature Logs (Thermistor String)	✓	✓
SCAMP (turbulent diffusion measurement)	✓	✓
Dye Tests (turbulent diffusion measurement)	✓	✓
Monthly Assemblage Grab Samples		✓
Water Quality Samples		✓



## Additional Sampling

### BAE Environmental Analysis Lab and CAAE Lab

- Chlorophyll a
- Total Density
- Total Keldahl Nitrogen
- Nitrate-Nitrite
- Ammonia
- Total Phosphorous
- Orthophosphate

### In Situ Measurements (Hydrolabs Sonde)

- Chl a, phycocyanin, DO, pH, Temp

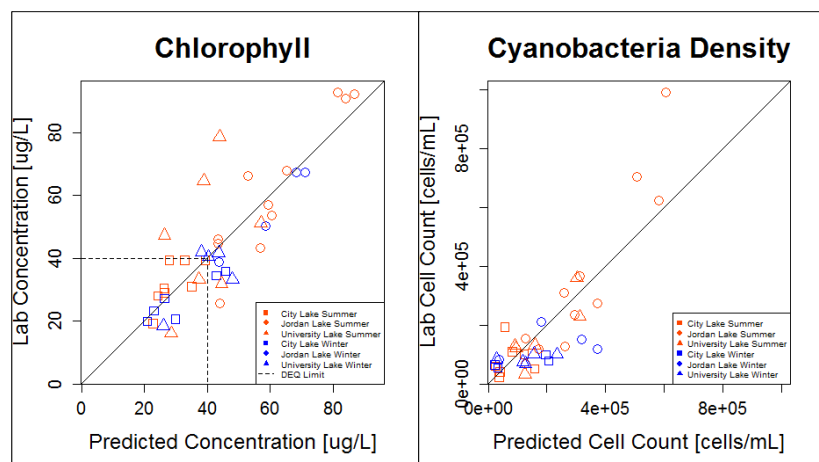
### Couple to Ongoing Monitoring Efforts

- Municipalities, DWR, EPA STORET

## Regression Modeling

$$\text{Bloom Metric} \sim C_{sf} + P_{sf}$$

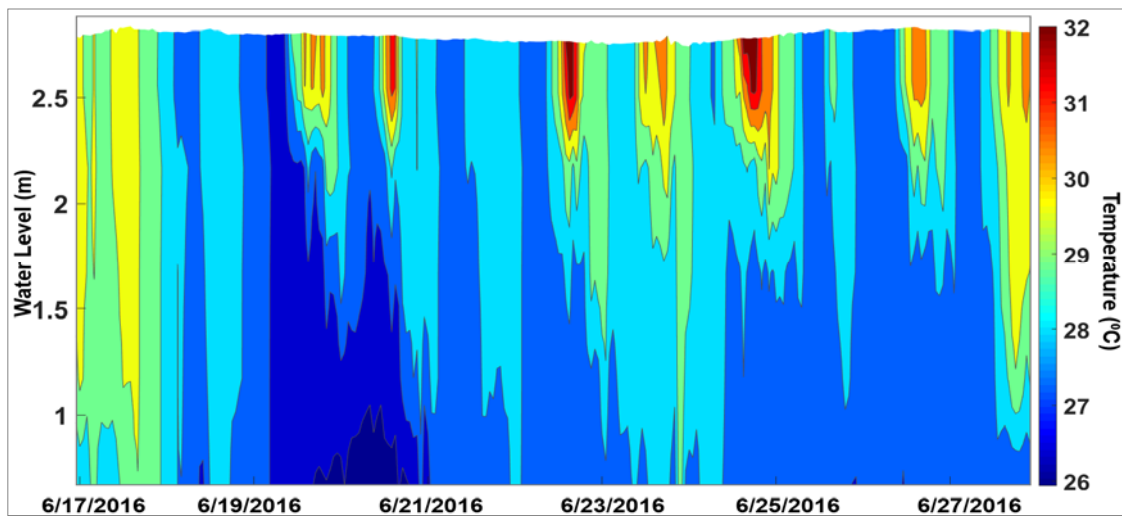
- $C_{sf}$  ~ Sonde Fluorescence of Chlorophyll  $\alpha$  [ $\mu\text{g/L}$ ]
- $P_{sf}$  ~ Sonde Fluorescence of Phycocyanin [cells/mL]
- Linear regression modeling was used to relate the sonde fluorescence data to the grab sampling data.
- This data will be used in a hierarchical modeling framework





## Mechanistic and Statistical Modeling (preliminary results)

Observed Temperature in Jordan Lake  
(thermistor chains)



## Mechanistic Model - Approximating Mixing with Thermistor Chain

We are developing a one dimensional hydrodynamics model for predicting the vertical distribution of temperature, density, and diffusion in lakes and reservoirs.

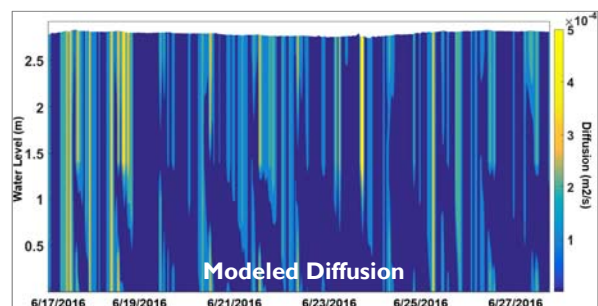
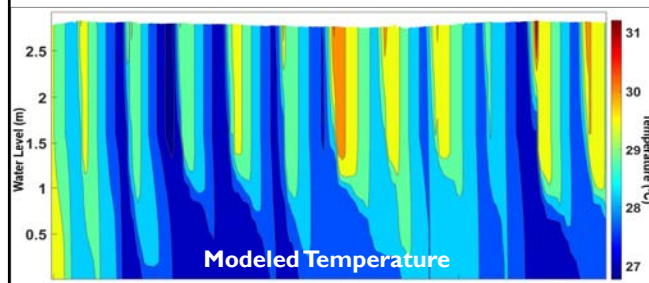
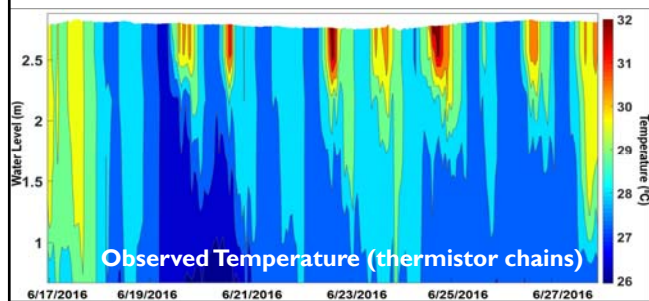
❖ Governing equation

$$\frac{\partial T}{\partial t} = \underbrace{\frac{1}{A(z)} \frac{\partial}{\partial z} \left\{ A(z) [\alpha + E(z, t)] \frac{\partial T}{\partial z} \right\}}_{\text{Diffusion}} + \underbrace{\frac{1}{A(z)} \frac{1}{C_p \rho} \frac{\partial [\Phi A(z)]}{\partial z}}_{\text{Meteorology}}$$

❖ Turbulent Diffusion, E

$$E(z, t) = f(Uw, \rho, z)$$

## Mathematical Heat Model



# Potential Relationships

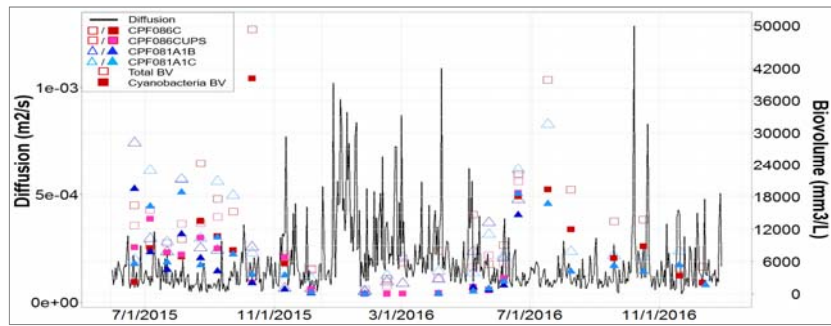


Figure 2(a): Plots of estimated diffusion rates with biovolume (includes total biovolume and cyanobacteria biovolume) with a two-year period from 2015 to 2016.

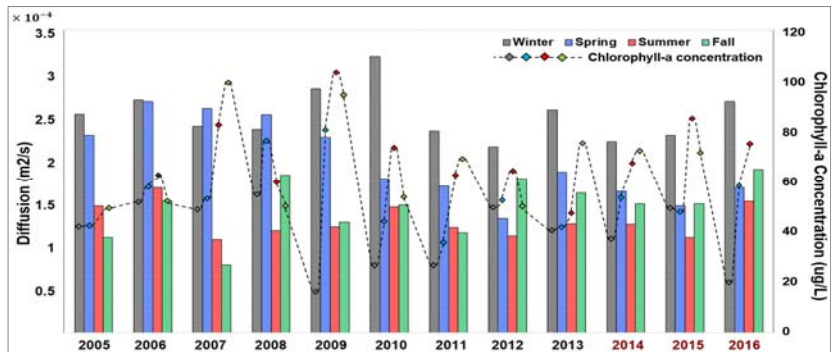


Figure 2(b): Averaged seasonal diffusion rates and chlorophyll-a concentration with a 12-year period.

## Multiple Linear Regression (early results)

NC STATE UNIVERSITY

- Model mainly based on following predictor variables with a prediction of Chlorophyll-a concentration

$$Chla \sim \log(Q) + \log(E) + \log(J_{sn})$$

Chla – Chlorophyll-a Concentration (ug/L);  
 Q – Flowrate (cfs);  
 E – Diffusion ( $m^2/s$ );  
 $J_{sn}$  – Solar Radiation ( $W/m^2$ )

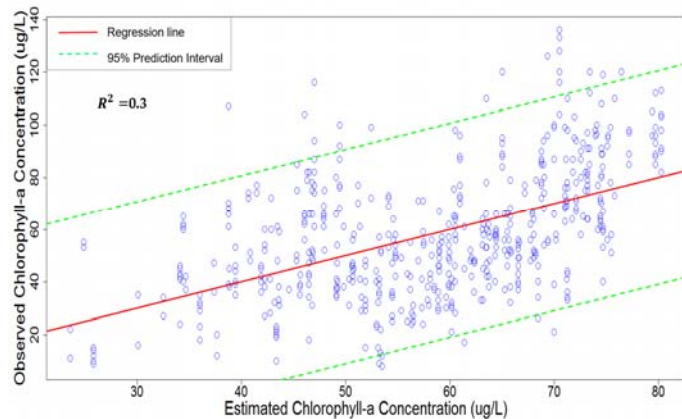


Table 1: Sign of the parameters and significance for the variables.

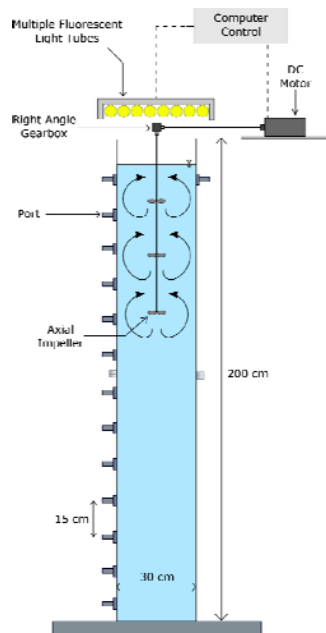
	Intercept	Log(Q)	Log(E)	Log( $J_{sn}$ )
Averaging Period (day)	NA	30	30	10
Sign of Trend Coeff.	-	-	-	+
Significance	***	***	***	***

Note: Significant codes: 0 \*\*\*\*\*

## On the horizon

- Continued monitoring
- Additional tweaking of mechanistic model
- Deeper analysis with statistical modelling
- Experimental configuration to study water column dynamics...

## The Water Column Reactor



NC State Seed Grant in collaboration with Dan Obenour and Astrid Schnetzer

### Components:

1. Reactor design (13 ports)
2. Mixing (axial impellers)
3. Light (fluorescents/LEDs)
4. Biology (cultured *Microcystis*)



## The People



## The Municipalities/Organizations/People

- NC DEQ: Water Resources – Jason Green and crew
- NSF
- NC WRRI and Urban Water Consortium
- NC State Office of Research Innovation and Economic Development (ORIED)
- OWASA
- City of High Point
- US Army Corps
- Bill Frazier

## Questions?

Tarek N. Aziz  
[tnaziz@ncsu.edu](mailto:tnaziz@ncsu.edu)

## Dye Test Model

$$\frac{\partial C}{\partial t} = E * \frac{\partial^2 C}{\partial z^2}$$

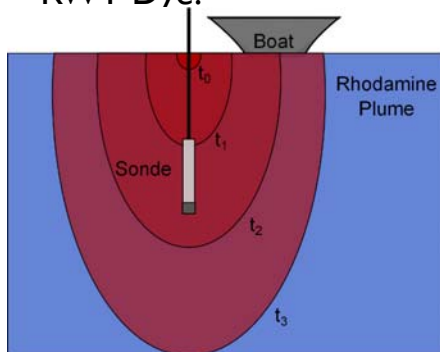
- E = Diffusivity (m<sup>2</sup>/s)
- C = Concentration of Rhodamine WT Dye (g/m<sup>3</sup>)
- t = time (sec) ; z = depth (m)

$$C_i^{n+1} = C_i^n + \frac{E * \Delta t}{\Delta z^2} (C_{i+1}^n - 2C_i^n + C_{i-1}^n)$$

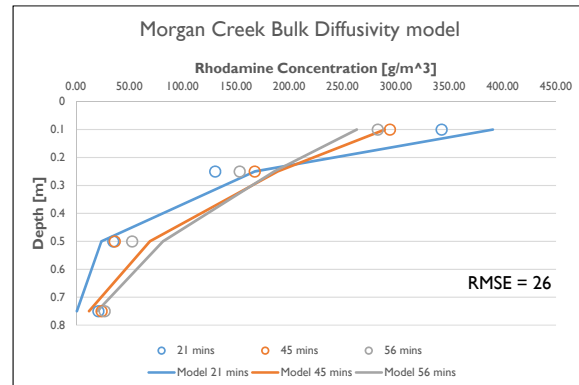
- Forward-time, center-space finite differencing model for time “n” at depth “i”

## Dye Test Modeling

- Dye tests on Jordan Lake with RWT Dye.



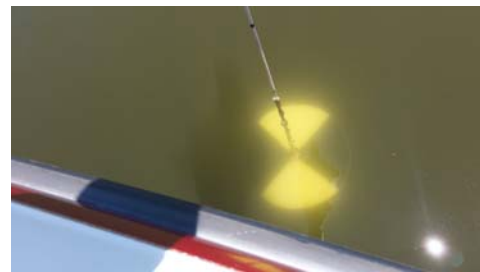
Schematic of the Dye Release Test



Results of the most recent dye test. Diffusivity was found to be **0.16 cm<sup>2</sup>/s**

## Multi-year Monitoring Effort

- Summer 2015 (Jordan Lake):
  - Monthly in-situ physical, chemical, and biological measurements
    - Temperature, LDO, pH, Conductivity, Chlorophyll Fluorescence, Phycocyanin Fluorescence
  - Hourly temperature logs from thermistor strings
  - SCAMP deployments
  - Dye tests
- Since Summer 2016:
  - Jordan Lake, University Lake, High Point Lake
  - Monthly in-situ physical, chemical, and biological measurements
    - Temperature, LDO, pH, Conductivity, Chlorophyll Fluorescence, Phycocyanin Fluorescence
  - Monthly phytoplankton assemblage and extracted chlorophyll grab samplings.
  - Hourly temperature logs from thermistor strings
  - SCAMP deployments
  - Dye Tests



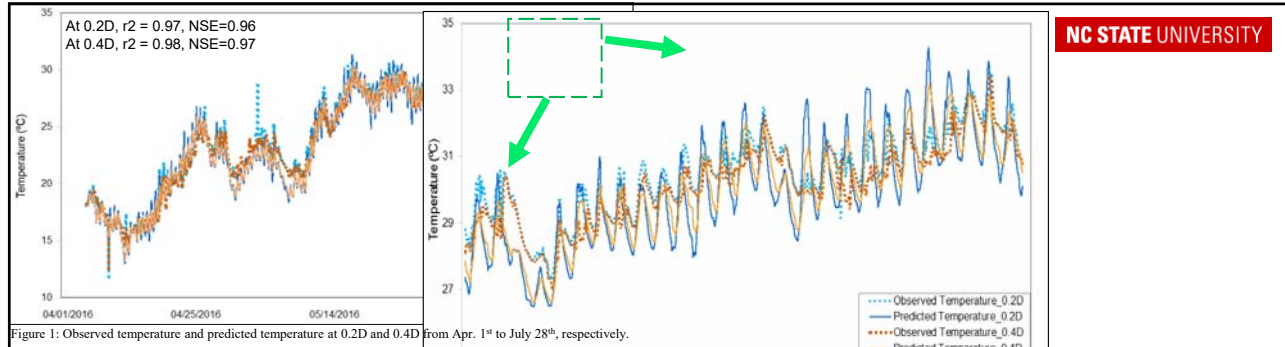


Figure 1: Observed temperature and predicted temperature at 0.2D and 0.4D from Apr. 1<sup>st</sup> to July 28<sup>th</sup>, respectively.

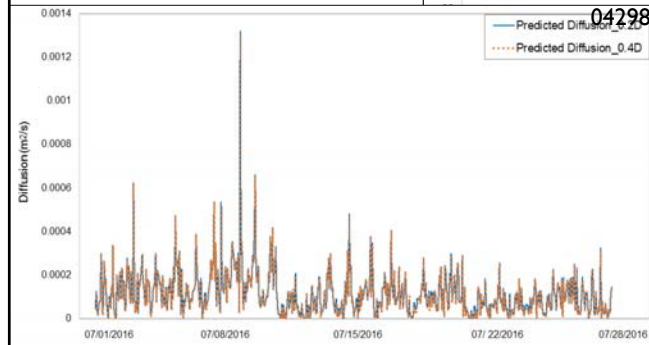


Figure 3: Predicted diffusion at 0.2D and 0.4D from July 1<sup>st</sup> to July 28<sup>th</sup>, respectively.

Figure 2: Observed temperature and predicted temperature at 0.2D and 0.4D from July 1<sup>st</sup> to July 28<sup>th</sup>, respectively.

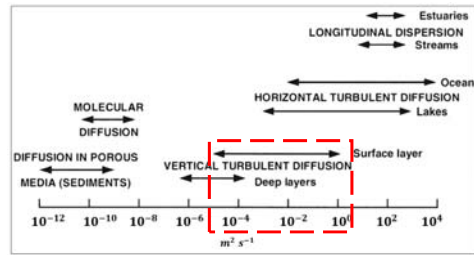


Figure 4: Typical ranges of the diffusion coefficient in natural waters and sediments (Chapra, 2008).